Total Partial Synthesis of (3S, 6S)-(+)-3,7-Dimethyl-6-hydroxy-3-acetoxyocta-1,7-diene and (3S, 6S)-(-)-3,7-Dimethylocta-1,7-diene-3,6-diol from Geraniol[†]

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The synthesis of (3S, 6S)-(+)-3,7-dimethyl-6-hydroxy-3-acetoxyocta-1,7-diene and (3S, 6S)-(-)-3,7-dimethylocta-1,7-diene-3,6-diol *via* the rearrangement of the chiral 2,3-epoxy alcohol, with the system Ph₃P, pyridine, I₂ and H₂O, is described.

Compound 1,¹ a novel monoterpenoid, has been isolated from *Mutisia spinosa* (Compositae) and compound **2** has been isolated from *Cinnamonum camphora*.² The absolute configurations at C-3 and C-6 of **1** and **2** have not been determined. Photooxidation of linally acetate and linalool⁴ gave **1** and **2**, respectively. However the authors obtained an isomeric mixture. To our knowledge, the asymmetric synthesis of **1** and **2** has not been reported. Herein we report the total synthesis of (3*S*, 6*S*)-(+)-**1** and (3*S*, 6*S*)-(-)-**2** from geraniol **3** (Scheme 1).



 K_2CO_3 , MeOH, r.t., 2 h, 96%.

Geraniol **3** was epoxized by Sharpless asymmetric epoxidation [L-(+)-DET was employed].⁵ By our method,⁶ rearrangement of epoxide **4** gave linalool **5** in 98% yield and > 95% e.e.⁷ Epoxide **4** was iodinated with Ph₃P (3 equiv), pyridine (4 equiv) and I₂ (1.5 equiv) in dry diethyl ether and CH₃CN, then the iodide was converted to linalool **5** by addition of 1 equiv. H₂O.

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[†] This is a **Short Paper** as defined in the Instructions for Authors, Section 5.0 [see *J. Chem. Research* (S), 1999, Issue 1]; there is therefore no corresponding material in *J. Chem. Research* (M).

Linalool **5** was protected with Ac₂O, Et₃N and DMAP,⁸ then SeO₂ oxidation gave the allylic alcohol **6**.⁹ Sharpless asymmetric epoxidation⁵ of the allylic alcohol **5** with L-(+)-DET led to the (6*S*, 7*S*)- and (6*R*, 7*R*)-epoxy alcohols in a ratio of *ca*. 10:1. Rearrangement of epoxy alcohol **7** gave the title compound **1** in 90% yield. Treatment of **1** with K₂CO₃ in MeOH give the diol **2** in 96% yield. As the authors did not report the optical rotation of natural **1** and **2**, we could not determine the absolute configuration of C-3 and C-6.

Experimental

IR spectra were recorded on a Nicolet FT 170SX as liquid films, ¹H NMR spectra were recorded on a Bruker AC-80 or Bruker AM-400 spectrometer and ¹³C NMR spectra were recorded on a 100 MHz spectrometer in CDCl₃ solution using TMS as internal reference. IR spectra were obtained using a FT 170SX (film) spectrophotometer. Mass spectra were measured on a VG ZAB-HS spectrometer by direct inlet at 70 eV, and signals are given in *m*/*z* with relative intensity (%) in parentheses. Optical rotation measurements were carried out on a Jasco 20C polarimeter.

Rearrangement of Compound 4 to 5.- To a stirred solution of epoxy alcohol 1 (500 mg, 2.94 mmol) in dry Et₂O-CH₃CN (5:3, 16 mL) was added sequentially Ph₃P (2.31 g, 8.82 mmol), pyridine (0.94 mL, 11.76 mmol), I_2 (1.12 g, 4.41 mmol) at 0 °C. After stirring for 2h at 0 °C, H₂O (53 $\mu L,$ 2.94 mmol) was added to the system. The reaction mixture was refluxed for 10 h at 38 °C, then 20% Na₂S₂O₃ (aq) (2mL) and saturated NaHCO₃ (aq) (2mL) were added to quench the reaction and the organic layer extracted with ether $(3 \times 50 \text{ mL})$. The combined ether extracts were washed with 5% HCl ($4 \times 10 \text{ mL}$), saturated NaHCO₃ (10 mL), H₂O and brine, then dried. Evaporation of the solvent gave the residue, which was flash chromatographed eluting with light petroleum-ethyl acetate (6:1 v/v) to afford 4 (445 mg, 98%) as a colorless oil. $[\alpha]_{20}^{20}$ + 21.2 (c 0.6, CHCl₃); $\delta_{\rm H}(400$ MHz) (CDCl₃): 5.91 (1H, dd, J = 17.3, 10.7 Hz, CH=), 5.21 (1H, d, J = 17.3 Hz,CH=), 5.11 (1H, t J = 6.7 Hz CH=), 5.07 (1H, d, J = 10.7 Hz, CH=), 2.08-1.94 (2H, m, CH₂), 1.68 (3H, s, CH₃), 1.65 (3H, s, CH₃), 1.58-1.50 (2H, m, CH₂), 1.28 (3H, s, CH₃).

Acetylation and SeO2 Oxidation of Linalool 5.- A mixture of linalool (445 mg, 2.9 mmol), triethylamine (0.61 mL, 4.35 mmol), acetic anhydride (0.41 mL, 4.35 mmol) and DMAP (28 mg, 0.23 mmol) was stirred at r.t. for 10 h. The mixture was extracted with ether $(3 \times 50 \text{ mL})$ and the organic layer was washed with 5% HCl $(4 \times 5 \text{ mL})$, saturated NaHCO₃ (10 mL), H₂O and brine, then dried. Evaporation of the solvent gave the residue, which was flash chromatographed eluting with light petroleum-ethyl acetate (15:1 v/v) to afford linalyl acetate (400 mg, 70%) as a colorless oil. A solution of linalyl acetate (400 mg, 2.04 mmol) in 5 mL CH₂Cl₂ was added to a solution of SeO₂ (113 mg, 1.02 mmol) and Bu^tOOH (70%, 1.13 mL, 4.08 mmol) in 15 mL CH_2Cl_2 and the reaction mixture was stirred at r.t. for 2 h. The mixture was extracted with ether $(3 \times 50 \text{ mL})$ and the organic layer washed with 10% KOH ($4 \times 5 \text{ mL}$), saturated NaHCO₃ (10 mL), H₂O and brine, then dried. Evaporation of the solvent gave the residue which was flash chromatographed eluting with light petroleum-ethyl acetate (6:1 v/v) to afford linalyl acetate (210 mg, 51%) as a colorless oil. $[\alpha]_D^{21} + 8.61$ (*c* 2.4, CHCl₃); $\delta_H(80 \text{ MHz})$ (CDCl₃): 5.90 (dd, J = 17.0, 10.8 Hz, 1H, CH=), 5.30 (t, J = 6.4 Hz, 1H, CH=), 4.00 (s, 2H, CH₂), 2.02 (s, 3H, CH₃), 2.18–1.92 (m, 2H, CH₂), 1.74–1.59 (m, 2H, CH₂), 1.65 (s, 3H, CH₃), 1.28 (s, 3H, CH₃).

Epoxidation of 6 to 7.—To a mixture of $Ti(OP_4)_4$) (0.3 mL, 1 mmol), CaH₂ (15 mg), 4 Å molecular sieves (40 mg) and silica gel H (20 mg) in 8 mL dry CH₂Cl₂ was added L-(+)-DET (0.21 mL, 1.2 mmol) under Ar at -20 °C. After 10 min, the allylic alcohol 6 (210 mg, 1 mmol) was added. The reaction mixture was stirred for another 10 min and Bu^tOOH (1.1mL, 3.16 M in toluene, 2 mmol) was added at -40 °C. After 8 h at -20 °C, 2 mL 10% aqueous tartaric acid was added. After stirring for 1h, the mixture was extracted with ether $(3 \times 50 \text{ mL})$ and the organic layer was washed with 10% KOH ($2 \times 10 \text{ mL}$), saturated NaHCO₃ (10 mL), H₂O and brine, then dried. Evaporation of the solvent gave the residue, which was flash chromatographed eluting with light petroleum–ethyl acetate (4:1 v/v) to afford epoxy alcohol 7 (177 mg, 77%) as a colorless oil. $[\alpha]_D^{24} - 6.4$ (c 0.75, CHCl₃). IR: 3454, 3089, 2974, 2934, 1736, 1643, 1247, 1027, 927, 875 cm⁻¹. EIMS: *m/z* 197(0.2%, M – 31), 153 (0.4), 137(2), 111(6), 94(20), 79(27), 71(30), 55(24), 43(100). $\delta_{\rm H}$ (400 MHz, CDCl₃): 5.91 (dd, J = 17.3 10.8 Hz, 1H, CH \Rightarrow , 5.15 (d, J = 17.3 Hz, 1H, CH \Rightarrow , 5.11 (d, J = 10.8 Hz, 1H, CH), 3.66 (d, J = 12.1 Hz, 1H, OCH), 3.54 (d, J = 12.1 Hz, OCH) 3.01 (t, J = 6.4 Hz, 1H epoxy H), 2.00 (s, 3H, CH₃), 2.03–1.95 (m, 1H, CH), 1.91-1.84 (m, H, CH), 1.65-1.59 (m, 1H, CH), 1.56-1.47 (m, 1H, CH), 1.55 (s, 3H, CH₃), 1.30 (s, 3H, CH₃).

(+)-3,7-Dimethyl-6-hydroxy-3-acetoxyocta-1,7-diene (1).-To a solution of epoxy alcohol 7 (140 mg, 0.61 mmol) in dry Et₂O-CH₃CN (5:3, 8 mL) was added sequentially Ph₃P (482 mg, 3 mmol), pyridine (0.2 mL, 2.45 mmol) and I_2 (234 mg, 0.92 mmol) at 0 °C. After stirring for 2 h at 0 °C, H2O (11 µL, 0.61 mmol) was added to the system. The reaction mixture was refluxed for 6 h at 38 °C, then 20% Na₂S₂O₃ (aq) (2 mL) and saturated NaHCO₃ (aq) (2mL) were added to quench the reaction and the organic layer extracted with ether $(3 \times 50 \text{ mL})$. The combined ether extracts were washed with 5% HCl ($4 \times 10 \text{ mL}$), saturated NaHCO₃ (10 mL), H₂O and brine, then dried. Evaporation of the solvent gave the residue, which was flash chromatographed eluting with light petroleum-ethyl acetate (6:1 v/v) to afford 1 (118 mg, 91%) as a colorless oil. $[\alpha]_D^{24}$ + 6.0 (c 0.80, CHCl₃). IR: 3435, 3075, 2974, 2941, 1736, 1647, 1251, 1020, 901 cm⁻¹. EIMS: m/z 152(0.4%, M – HOAc), 137(4), 109(5), 93(7), 71(58), 67(65), 55(23), 43(100). $\delta_{\rm H}$ (400 MHz, CDCl₃): 5.93 (dd, J = 17.4, 10.9 Hz, 1H, CH=), 5.14 (d, J = 17.4 Hz, 1H, CH=), 5.11 (d, J = 10.9 Hz, 1H, CH=), 4.93, 4.84 (s, 2H, $CH_2 =$), 4.03 (t, J = 6.4 Hz 1H, CHO), 2.00 (s, 3H, CH_3 ,

2.05–1.92 (m, 2H, CH₂), 1.82–1.77 (m, 2H, CH₂), 1.70 (s, 3H, CH₃), 1.53 (s, 3H, CH₃), $\delta_{\rm C}(100$ MHz, CDCl₃): 169.93, 147.06, 141.51, 113.28, 111.29, 82.73, 75.77, 35.65, 28.68, 23.63, 22.33, 17.33.

(-)-3,7-*Dimethylocta*-1,7-*diene*-3,6-*diol* (2).—A mixture of **1** (15 mg, 0.064 mmol) in methanol (1mL) and anhydrous K_2CO_3 (8 mg) was stirred at r.t. for 2 h. The reaction mixture was diluted with Et₂O (30 mL), washed (2 × 5 mL H₂O, 5 mL brine), dried (anhydrous MgSO₄), and purified by column chromatography to yield **2** (12 mg, 96%). [α]_D²¹ – 7.53 (*c* 1.30, CHCL₃). IR: 3358, 3085, 2973, 2948, 2869, 1649, 1450, 1057, 999, 921, 899 cm⁻¹. EIMS: *m*/*z* 155(0.2%, M – 15), 137(4), 109(4), 96(4), 82(22), 71(61), 67(75), 55(39), 43(100), 41(56). $\delta_{\rm H}$ (400 MHz CDCl₃): 5.90 (dd, *J* = 17.2, 10.7 Hz, 1H, CH=), 5.22 (d, *J* = 17.2 Hz, 1H, CH=) 5.06 (d, *J* = 10.7 Hz, 1H, CH=), 4.95, 4.84 (s, 2H, CH₂=), 4.06 (t, *J* = 6.3 Hz, 1H, CHO), 2.17 (bs, 2H, 2OH), 1.71 (s, 3H, CH₃), 1.76–1.54 (m, 4H, 2CH₂), 1.29 (s, 3H, CH₃). $\delta_{\rm C}$ (100 MHz, CDCl₃) 147.34, 145.01, 111.76, 110.83, 75.60, 72.86, 37.66, 29.10, 27.88, 17.84.

This work was financially supported by the National Nature Science Foundation of China (Grant No. 29672015) and the Special Research Grant for Doctoral Sites in Chinese Universities.

Received, 22nd December 1998; Accepted, 28th January 1999 Paper E/8/07189F

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